Decoupled Sliding Mode Control for a Multivariable Nonlinear System

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ABSTRACT

This paper presents a decoupled control strategy using timevarying sliding surface-based Sliding Mode Controller (SMC) for a multivariable nonlinear system as an Ammonia Reactor system. The decoupled method provides a simple way to achieve asymptotic stability by dividing the system into three subsystems. Simulation results are presented for SMC comparing with a traditional PID controller. Then , the system is subjected to temperature disturbance to demonstrate the effectiveness and robustness of the controller.

General Terms

Decoupled Sliding Mode Control.

Keywords

Sliding Mode Controller; Ammonia Reactor; Decoupled system; PID controller.

1. INTRODUCTION

Sliding mode control provides (i) insensitivity to parameter variations; (ii) external disturbance rejection; and (iii) fast dynamic responses. These robust properties make this discontinuous control strategy very attractive [1-4].

The main idea of sliding mode control is to steer the system trajectories into the sliding surface and then keep them on it. So the control law consists of two parts: a continuous control which controls the system when its states are on the sliding plane and a discontinuous control which handles the uncertainties. In practice, the system states do not remain on the sliding surface after reaching it due to uncertainties in dynamical equations. Around the surface S, is often irritated by high frequency oscillations known as chattering [5-6].

The development of general sliding mode design approaches for multi-input nonlinear systems has been hampered by the need to reduce the nonlinear system to regular form [7-10]. Research on multivariable nonlinear systems with internal parameter uncertainties and external disturbances is important for both theoretical and practical reasons [11-13].

In this paper, a Decoupled Sliding Mode Controller (DSMC) design is used to control an Ammonia Reactor. The Reactor is divided into three subsystems with different switching surfaces. Comparing the DSMC controller with a traditional PID controller with exposing the system to temperature disturbance to demonstrate the effectiveness and robustness of the controller.

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2. SYSTEM DESCRIPTION

Ammonia synthesis is the first stage in the manufacturing of nitrous chemical products. In order to determine the optimum working regimes. This is determined by:

a. The direct synthesis reaction N2 + H2 \leftrightarrow NH3 is a reversible exothermic reaction which must take place at equilibrium to get a maximum conversion rate.

b. The conversion rate depends on: temperature (the temperature increase leads to ammonia decomposition, thus leading to conversion rate decrease), catalyst activity and pressure.

c. The synthesis columns are built with multiple fixed catalyst layers and Quenches to reduce the temperature and ammonia concentration at the input into the next layer

d. The catalyst is spent non-uniform, starting with the first layer.

3. AMMONIA REACTOR

The ammonia reactor is the heart of the synthesis plant. Ammonia reactor with a fixed-bed quench cooling is shown in Figure 1.

Where

T_F: Feed Gases Temperature.

 T_i : Inlet Gases Temperature.

T_o: Outlet Gases Temperature.

The reactor shown in Figure1 is a fixed bed reactor with fresh feed quenching between the beds.

The converter consists of three parts:

(a) The catalyst bed sections.

(b) The Quench sections.

(c) The heat-exchanger section.



Fig 1: Ammonia Reactor with Bed and Quenching Sections

The inlet gaseous mixture is split into four separate streams as:

(a) The main stream called the heat exchanger Flow (Q4).

(b) The second stream which enters 1st Quench (Q1).

(c) The third stream which enters 2nd Quench (Q2).

(d) The Fourth stream which enters 3rd Quench (Q3)[14].

4. MATHEMATICAL MODEL

4.1 Reactor Mathematical Model

The mathematical model of the ammonia reactor consists of one model for each bed. Each bed is described by a mass and energy balance, and is discretized into ten segments [14].

$$Q\frac{\partial c}{\partial z} = m_c r(T,c) \tag{1}$$

$$m_{c}C_{pc}\frac{\partial T}{\partial t} + QC_{pg}\frac{\partial T}{\partial z} = (-\Delta H_{rx})m_{c}r(T,c) + \Gamma m_{c}C_{pc}\frac{\partial^{2}T}{\partial z^{2}}$$
(2)

Where

t	Time	[sec.]
Z	Position in reactor.	
Т	Particle temperature	[K]
c	Concentration of ammonia	[kg NH3/kg gas]
$-\Delta$ Hrx	Heat of reaction	[J/kg.NH3]
C _{pc}	Heat capacity of catalyst	[J/kg cat.K]
C _{pg}	Heat capacity of gas	[J/kg.K]
m _c	Catalyst mass in the bed	[Kg]
Q	Gas flow through the bed	[Kg/s]
r(T, c)	Reaction rate	[kg.NH3/kg cat.sec.]
Γ	Dispersion coefficient	[1/sec]

The model may be discretized in space. So, the model can be rewritten in discretized equations for cell no. j as :

$$0 = Q(c_{j-1} - c_j) + m_{c,j}r(T_j, c_j)$$
(3)
$$m_{c,j}C_{pc}\frac{\partial T_i}{\partial t} = QC_{pg}(T_{j-1} - T_j) + (-\Delta H_{rx})m_{c,j}r(T_j, c_j)$$
(4)

Hence each bed is descretized into 10 cells, so 30 equations describe the reactor model.

4.2 The Quench Model

Fresh feed quenching of the ammonia concentration are entering the reactor between each bed. Quenches are modeled as follows:

$$T_{mix} = \frac{Q_b}{Q_b + Q_q} T_b + \frac{Q_q}{Q_b + Q_q} T_q$$
⁽⁵⁾

$$c_{mix} = \frac{Q_b}{Q_b + Q_q} c_b + \frac{Q_q}{Q_b + Q_q} c_q \tag{6}$$

The subscript q, b and mix refer to the quench stream. The reactor flow before the quench and the reactor flow after the quench, respectively. Temperature, concentration of ammonia and mass flow are denoted T, c and Q.

4.3 The Heat Exchanger Model

The model of the heat exchanger as follows:

$$T_i = \mathcal{C} T_0 + (1 - \mathcal{C}) T_F \tag{7}$$

Where ε is the heat exchanger efficiency

5. OPEN LOOP SIMULATUIN RESULTS From the reactor construction, inputs and outputs can be summarized as:

Reactor Inputs : The split Mass Flows enter the reactor (Q_1,Q_2,Q_3,Q_4) , Feed temperature of gasses. (T_F) , Feed ammonia mass fraction (C_F) and Feed pressure (P).

Reactor Outputs : Outlet gasses temperature (T_o) and outlet ammonia mass fraction (C_o) .



Fig 2: Reactor Outlet Temperature and Concentration.

The system is stable and the temperature was constant at 506.4°C. The resulted ammonia concentrations 29.3%. The peak in ammonia concentration at the beginning of reactor work, due to the transient temperatures state. Hence, the temperatures inside the beds and through the quenches not

reach the nominal value for beginning of conversion, so, a transient ammonia concentration is obtained. Till the temperature is stabilized and a stable reaction rate was obtained. The experts are not worry about the peak because of they prefer high outlet ammonia concentrations as commercial considerations.

6. FEED TEMPERATURE DISTURBANCE

The feed temperature has been disturbed to 210° C, i.e., for about 30% from the regular Feed temperature. The outlet temperature and ammonia concentration due to the temperature disturbance are shown in Figure 3.



Fig 3: Disturbed Outlet Temperature and concentration.

Figure 3 shows high temperature oscillations after the temperature has been reduced to 210°C. These vibrations resumed for about 130 minutes then, the system stabilizes at 238.7°C. This means that the temperature is reduced by about 53% from its steady state stable value. High oscillations similar to the temperature and the system stabilizes at

ammonia concentration about 4.78% . i.e., the concentration reduced to about 83.75% from its steady state stable value.

7. AMMONIA REACTOR CONTROLLER DESIGN

Some issues have to be mentioned before embarking on the control of the ammonia reactor.

Firstly, the ammonia reactor operating point possesses the highest temperature, which leads to the highest conversion of ammonia, although it results in a lower yield of ammonia [15-17].

Secondly, the oscillations given in Figure 3 appeared as a result of too low inlet temperature through the ammonia reactor. To avoid the instability, one can naturally increase the inlet temperature, T_F , or the reactor pressure (P), which may be no suitable for the ammonia production unit. Another possibility is to increase the heat recovery by reducing the flows through the other quenches (Q₁, Q₂, Q₃) so that more of the feed is preheated [18-19]. The reactor system could be stabilized using the mass flow trough the quenches to control the temperatures at the inlet to the beds. This means using the temperature feed mass flow Q₁, Q₂ and Q₃ as manipulated variables and the reactor inlet temperatures (after the quenches) as measurements. The block diagram of the controller as shown in Figure 4.

 T_{b1ref} , T_{b2ref} and T_{b3ref} : The Reference descretized Bed cells

	temperatures.
T_{b1} , T_{b2} and T_{b2}	: The Real descretized Bed cells
	temperatures.
u_1 , u_2 and u_3	: The Valves control signals.
Q_1, Q_2 and Q_3	: The mass flows enter the Quenches.
T _{out}	: The Outlet Reactor temperature
C _{out}	: The Outlet ammonia mass
	concentration.



Fig 4: Block Diagram of Ammonia Reactor Controller

The control signal can vary as :

$$0 \le u \le 1$$

where

(8)

u=1 the valve is fully opened and u=0 the valve is fully closed.

The main idea of the controller is to control the temperatures at the inlet to the beds. This is done by comparing the reference quenching outlet temperatures (T_{q}) . Hence, the controller produces the control signals (u) to control the mass flow that enter the three quenches of the reactor to satisfy the temperature required for the operation.

For verifying the effectiveness of the controller, some constraints are added to the valves. The quantity of flow gasses Q_1 , Q_2 and Q_3 through the quench valves should be at least 50% of the normal gasses, that can be flowed, i.e., $u \ge 0.5$

An increasing in ammonia concentration is expectable. Due to the remaining Feed Gas quantities, those were not passed through the valves V_1 , V_2 and V_3 (according to the control signal u_1 , u_2 and u_3). These remaining quantities are added to the main stream (see Figure 1), which increase the outlet ammonia concentration.

8. PID CONTROL

PID controller is used to control the quench valves by computing the errors due to the operation as shown: $e_{1} = T$, $e_{2} = T$.

$$e_{1} = T_{q1ref} - T_{q1}$$

$$e_{2} = T_{q2ref} - T_{q2}$$

$$e_{3} = T_{q3ref} - T_{q3}$$
(9)

Then computing the control signals by:

$$u_{1} = K_{c1} \left(e_{1} + \frac{1}{T_{I1}} \int e_{1} dt + T_{D1} \frac{de_{1}}{dt} \right)$$

$$u_{2} = K_{c2} \left(e_{2} + \frac{1}{T_{I2}} \int e_{2} dt + T_{D2} \frac{de_{2}}{dt} \right)$$

$$u_{3} = K_{c3} \left(e_{3} + \frac{1}{T_{I3}} \int e_{3} dt + T_{D3} \frac{de_{3}}{dt} \right)$$
(10)

Where K_{c1}, K_{c2} and K_{c3}

 T_{I1} , T_{I2} and T_{I3}

: Proportional Gains. : Integral Times.

 T_{D1} , T_{D2} and T_{D3} : Derivative Times.

The PID controller model is shown in Figure 5.



Fig 5: The PID controller model

Constant C ensures that $u \ge 0.5$. Simulation results are shown in Figure 6.





Fig 6: Disturbed Reactor Outlet Temperature and concentration using PID controller.

When the system is subjected to disturbance due to reducing the feed temperature, the reactor suffers from temperature oscillations. The PID controller tries to overcome the reduction in feed temperature by adjusting the quench valves. The system has temperature oscillations for about 105 minutes. The temperature reaches its min. value at 272.2°C then stabilizes at 500°C. So the temperature drop was about 45.56% from the normal value.

Also, high oscillations in the ammonia outlet concentration, which reach their min. value at 12.19%. Then, after 105 minutes, the system stabilizes with 30.29% ammonia concentrations.

From Figure 6 it is clear that, the system suffers from high oscillation for about 105 minutes, which is along time taken by the controller to overcome the disturbance.

9. VARIABLE STRUCTURE CONTROL FOR MULTIVARAIBLE SYSTEM

Consider a multivariable system described by the state equation:

$$x = Ax + Bu$$

state variables

$$y = Lx \tag{11}$$
 where

x(t) is the state vector, u(t) is the input vector, y(t) is the output vector and A,B,L are constant matrices. The sliding surface is chosen in error domain as:

$$S = C^T e \tag{12}$$

where C is a parameter matrix. The error is defined as
$$e = x_d - x$$
(12)

 $e^{-x_d - x}$ (13) where xd is the desired state. The error derivatives are the

$$e_i = e_{i+1}$$
 where $i = 1, ..., n-1$

$$e_n = a_1 e_1 + a_2 e_2 + \dots + a_n e_n + u(t) + d(t)$$
(14)

Hence d(t) represents the external disturbance. The controller output signal u can be divided into an equivalent control input

 u_{eq} and a hitting control input uh, it has the following control law:

$$u = u_{eq} + u_h$$
(15)
With Sliding Condition

$$\lim_{s \to 0^+} \dot{s} < 0, \qquad \lim_{s \to 0^-} \dot{s} > 0 \tag{16}$$

At S(x) = S(x) = 0 is called a switching surface or switching hyperplane as in Figure 7.



Fig 7: The Sliding surface (S)

The Equivalent control u_{eq} is given by:

$$S = 0 = c_{1}e_{1} + c_{2}e_{2} + \dots + c_{n-1}e_{n-1} + e_{n} \quad \text{where} \quad c_{n} = 1$$

$$S = C^{T} e(t) = 0$$

$$c_{1}e_{1} + c_{2}e_{2} + \dots + c_{n-1}e_{n-1} + e_{n} = 0$$

$$a_{1}e_{1} + (a_{2} + c_{1})e_{2} + \dots + (a_{n} + c_{n-1})e_{n} + u_{eq} + d(t) = 0$$

$$u_{eq}(t) = -a_{1}e_{1} - (a_{2} + c_{1})e_{2} - \dots - (a_{n} + c_{n-1})e_{n} - d(t)$$

So the control input u can be expressed as
(17)

 $u(t) = -a_1e_1 - (a_2 + c_1)e_2 - \dots - (a_n + c_{n-1})e_n - d(t) + u_h$ (18)

10. DECOMPOSITION OF MULTIVARIABLE SYSTEM INTO SUBSYSTYEMS

Since it is required to produce a control input u(t) that will be applied to the system. To reduce the complexity of the system, the ammonia reactor system can be decomposed into three subsystems (see Appendix B), each subsystem contains single Quench and single Bed this can be as shown in Figure 8.





Each subsystem has its own sliding mode controller with the following sliding equations:

$$S_{1} = \sum_{i=1}^{n} c_{i}e_{i} \quad and \quad c_{n} = 1$$

$$S_{2} = \sum_{j=1}^{m} c_{j}e_{j} \quad and \quad c_{m} = 1$$

$$S_{3} = \sum_{k=1}^{r} c_{k}e_{k} \quad and \quad c_{r} = 1$$
(19)
where

$$\begin{aligned} e_i &= T_{ib1ref} - T_{ib1} \\ e_j &= T_{jb2ref} - T_{jb2} \end{aligned}$$
(20)

$$e_{\nu} = T_{\nu h 3 ref} - T_{\nu h 3}$$

So SMC Controller uses control low is expressed as $u = u_{eq} + Msat(S)$

(21)where M is positive constant.

Figure 9 shows the contents of the Sliding Mode Controller for each subsystem. The flow quantity through each valve is constrained to be between 50% and 100% from the normal value.



Fig 9: DSMC Controller.

11. SIMULATION RESULTS

In this section, the validity and effectiveness of the proposed controller scheme are examined through the fast recovery and the stability of the system. Compare these results with traditional PID controller.

Figure 10 shows a stable behavior of the reactor regardless 50% constrains and the temperature disturbance. When a sudden change in the feeding temperature at time 135mins. There are little ripples in the mass concentration due to the temperature disturbance. The outlet mass concentration is about 31%. But, when applying the disturbance there is some little oscillations for 5 mins. Approximately, then the system stabilized after 26.5 mins.

Table 1 summarizes the results of the controllers

 Table 1. Controllers Comparison

System Compare	PID	DSMC
Oscillations	Very high	little
Time of Recovery (min)	105	26.5
Ammonia Concentration	30.29%	31%
Ammonia Quantity(Tons/H)	76.33	78.12

The table shows that DSMC has achieved the fast recovery time with little ripples. It takes 26.5 min to overcome the disturbance. While PID controller has the highest ripples with largest time of recovery.





1g 10: Disturbed Reactor Outlet Temperature and Concentrations using DSMC controller.

12. CONCLUSION

This paper presents the design of control system for controlling an Ammonia Reactor using the decoupled sliding mode control algorithm. The decoupled technique divides the system into three subsystems each contains a Quench and a Bed which can be separately modeled and controlled. Subject the reactor to temperature disturbance with adding constraints to the quantity of flow gasses to verify the effectiveness of the controller. Comparing the controller with traditional PID controller which achieves high oscillation in outlet ammonia concentration, illustrates the robustness of the suggested controller.

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Appendix A : Data Model

Gas heat capacity, CPg	3500	J/kg, K
Catalyst Heat capacity, CPc	1100	J/kg, K
Heat of reaction, - Δ Hrx	2.7x106	J/kg NH3
Volume, bed 1	6.69	m3
Volume, bed 2	9.63	m3
Volume, bed 3	15.2	m3
Catalyst bulk density, pcat	2200	Kg/m3
Typical gas density	50	Kg/m3

58	Tons/h
53	Tons/h
32	Tons/h
127	Tons/h
0.0417	%
0.2396	%
0.7187	%
300	°C
170	Bar
	58 53 32 127 0.0417 0.2396 0.7187 300 170

Appendix B : Subsystems of Ammonia Reactor Model

Subsystem1: Which contains Quench1 and Bed1

x ₁		$-\alpha_1$	0	0	0	0	0	0	0	0	0	<i>x</i> ₁	
\dot{x}_2		α_1	$-\alpha_1$	0	0	0	0	0	0	0	0	x_2	
$\dot{x_3}$		0	α_1	$-\alpha_1$	0	0	0	0	0	0	0	<i>x</i> ₃	
$\dot{x_4}$		0	0	α_1	$-\alpha_1$	0	0	0	0	0	0	<i>x</i> ₄	
<i>x</i> ₅	=	0	0	0	α_1	$-\alpha_1$	0	0	0	0	0	x5	1
<i>x</i> ₆		0	0	0	0	α_1	$-\alpha_1$	0	0	0	0	x ₆	ľ
x ₇		0	0	0	0	0	α_1	$-\alpha_1$	0	0	0	x7	
<i>x</i> ₈		0	0	0	0	0	0	α_1	$-\alpha_1$	0	0	x ₈	
<i>x</i> ₉		0	0	0	0	0	0	0	α_1	$-\alpha_1$	0	<i>x</i> ₉	
<i>x</i> ₁₀		0	0	0	0	0	0	0	0	α_1	- α ₁	x ₁₀	ļ

$\beta_1 + (1-\varepsilon)\alpha_1$	$\delta_{\rm l}\varepsilon$	$\frac{-\Delta Hrx}{Cpc^*mc_1}(Q_1+Q_4)$	$\frac{-\Delta Hrx}{Cpc^*mc_1}C_1 = 0$	0 ($\frac{-\Delta Hrx}{Cpc^*mc_1}C_1$	
0	0	0	$\frac{-\Delta Hrx}{Cpc^*mc_1}(C_2 - C_1) = 0$	0	$\frac{-\Delta Hrx}{Cpc^*mc_1}(C_2 - C_1)$	
0	0	0	$\frac{-\Delta Hrx}{Cpc^*mc_1}(C_3-C_2) = 0$	0	$\frac{-\Delta Hrx}{Cpc^*mc_1}(C_3 - C_2)$	T_F
0	0	0	$\frac{-\Delta Hrx}{Cpc^*mc_1}(C_4 - C_3) = 0$	0	$\frac{-\Delta Hrx}{Cpc^*mc_1}(C_4 - C_3)$	T 30
0	0	0	$\frac{-\Delta Hrx}{Cpc^*mc_1}(C_5 - C_4) = 0$	0	$\frac{-\Delta Hrx}{Cpc^*mc_1}(C_5 - C_4)$	C_F O_1
0	0	0	$\frac{-\Delta Hrx}{Cpc^*mc_1}(C_6 - C_5) = 0$	0	$\frac{-\Delta Hrx}{Cpc^*mc_1}(C_6 - C_5)$	Q_2
0	0	0	$\frac{-\Delta Hrx}{Cpc^*mc_1}(C_7 - C_6) = 0$	0 0	$\frac{-\Delta Hrx}{Cpc^*mc_1}(C_7 - C_6)$	Q_3
0	0	0	$\frac{-\Delta Hrx}{Cpc^*mc_1}(C_8 - C_7) = 0$	0	$\frac{-\Delta Hrx}{Cpc^*mc_1}(C_8 - C_7)$	Q_4
0	0	0	$\frac{-\Delta Hrx}{Cpc^*mc_1}(C_9 - C_8) = 0$	0	$\frac{-\Delta Hrx}{Cpc^*mc_1}(C_9 - C_8)$	
0	0	0	$\frac{-\Delta Hrx}{Cpc^*mc_1}(C_{10}-C_9) = 0$	0 0	$\frac{-\Delta Hrx}{Cpc^*mc_1}(C_{10}-C_9)$	

Subsystem2: Which contains Quench2 and Bed2

$\begin{bmatrix} x_{11} \end{bmatrix}$	$-\alpha_2$	0	0	0	0	0	0	0	0	0	x ₁₁
x ₁₂	α_2	$-\alpha_2$	0	0	0	0	0	0	0	0	<i>x</i> ₁₂
x ₁₃	0	α_2	$-\alpha_2$	0	0	0	0	0	0	0	<i>x</i> ₁₃
x ₁₄	0	0	α_2	$-\alpha_2$	0	0	0	0	0	0	<i>x</i> ₁₄
x15	0	0	0	α_2	$-\alpha_2$	0	0	0	0	0	<i>x</i> ₁₅
x ₁₆	0	0	0	0	α_2	$-\alpha_2$	0	0	0	0	<i>x</i> ₁₆
x ₁₇	0	0	0	0	0	α_2	$-\alpha_2$	0	0	0	<i>x</i> ₁₇
x_{18}	0	0	0	0	0	0	α_2	$-\alpha_2$	0	0	<i>x</i> ₁₈
<i>x</i> ₁₉	0	0	0	0	0	0	0	α_2	$-\alpha_2$	0	<i>x</i> ₁₉
$[x_{20}]$	0	0	0	0	0	0	0	0	α_2	$-\alpha_2$	x20

β_2	δ_2	$-\frac{-\Delta Hrx}{Cpc^*mc_2}(Q_2)$	$-\frac{-\Delta Hrx}{Cpc^*mc_2}(Q$	$Q_1 + Q_4$) $\frac{-\Delta Hrx}{Cpc^*mc_2}C_{11}$	$\frac{-\Delta Hrx}{Cpc^*mc_2}C_{11}$	0	$\frac{-\Delta Hrx}{Cpc^*mc_2}C_{11}$	where
0	0	0	0	$\frac{-\Delta Hrx}{Cpc^{*}mc_{2}}(C_{12}-C_{11})$	$\frac{-\Delta Hrx}{Cpc^*mc_2}(C_{12}\!-\!C_{11})$	0	$\frac{-\Delta Hrx}{Cpc^*mc_2}(C_{12}-C_{11})$	
0	0	0	0	$\frac{-\Delta Hrx}{Cpc^{*}mc_{2}}(C_{13}-C_{12})$	$\frac{-\Delta Hrx}{Cpc^*mc_2}(C_{13}\!-\!C_{12})$	0	$\frac{-\Delta Hrx}{Cpc^{*}mc_{2}}(C_{13}-C_{12})$	
0	0	0	0	$\frac{-\Delta Hrx}{Cpc^{*}mc_{2}}(C_{14}-C_{13})$	$\frac{-\Delta Hrx}{Cpc^*mc_2}(C_{14}-C_{13})$	0	$\frac{-\Delta Hrx}{Cpc^{*}mc_{2}}(C_{14}-C_{13})$	T 10 C _F
0	0	0	0	$\frac{-\Delta Hrx}{Cpc^*mc_2}(C_{15}-C_{14})$	$\frac{-\Delta Hrx}{Cpc^{*}mc_{2}}(C_{15}-C_{14})$	0	$\frac{-\Delta Hrx}{Cpc^{*}mc_{2}}(C_{15}\!-\!C_{14})$	C ₁₀
0	0	0	0	$\frac{-\Delta Hrx}{Cpc^*mc_2}(C_{16}-C_{15})$	$\frac{-\Delta Hrx}{Cpc^*mc_2}(C_{16}-C_{15})$	0	$\frac{-\Delta Hrx}{Cpc^{*}mc_{2}}(C_{16}-C_{15})$	Q ₁ Q ₂
0	0	0	0	$\frac{-\Delta Hrx}{Cpc^*mc_2}(C_{17}-C_{16})$	$\frac{-\Delta Hrx}{Cpc^{*}mc_{2}}(C_{17}-C_{16})$	0	$\frac{-\Delta Hrx}{Cpc^{*}mc_{2}}(C_{17}-C_{16})$	Q ₃
0	0	0	0	$\frac{-\Delta Hrx}{Cpc^{*}mc_{2}}(C_{18}-C_{17})$	$\frac{-\Delta Hrx}{Cpc^{*}mc_{2}}(C_{18}-C_{17})$	0	$\frac{-\Delta Hrx}{Cpc^{*}mc_{2}}(C_{18}\!-\!C_{17})$	Q ₄
0	0	0	0	$\frac{-\Delta Hrx}{Cpc^*mc_2}(C_{19}-C_{18})$	$\frac{-\Delta Hrx}{Cpc^*mc_2}(C_{19}-C_{18})$	0	$\frac{-\Delta Hrx}{Cpc^{*}mc_{2}}(C_{19}-C_{18})$	
0	0	0	0	$\frac{-\Delta Hrx}{Cpc^{*}mc_{2}}(C_{20}-C_{19})$	$\frac{-\Delta Hrx}{Cpc^*mc_2}(C_{20}-C_{19})$	0	$\frac{-\Delta Hrx}{Cpc^*mc_2}(C_{20}-C_{19})$	

Subsystem3: Which contains Quench3 and Bed3

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x21	$-\alpha_3$	0	0	0	0	0	0	0	0	0	x21	
x22	α3	$-\alpha_3$	0	0	0	0	0	0	0	0	x22	
x23	0	α_3	$-\alpha_3$	0	0	0	0	0	0	0	x23	
x24	0	0	α_3	$-\alpha_3$	0	0	0	0	0	0	x24	
x25	0	0	0	α_3	$-\alpha_3$	0	0	0	0	0	x25	
x26	0	0	0	0	α_3	$-\alpha_3$	0	0	0	0	x26	+
x27	0	0	0	0	0	α_3	$-\alpha_3$	0	0	0	x27	
x28	0	0	0	0	0	0	α_3	$-\alpha_3$	0	0	x28	
x29	0	0	0	0	0	0	0	α_3	$-\alpha_3$	0	x29	
x ₃₀	0	0	0	0	0	0	0	0	α_3	$-\alpha_3$	x ₃₀	

β	δ_3	$-\frac{-\Delta Hrx}{Cpc^*mc_3}(Q_3)$	$-\frac{-\Delta Hrx}{Cpc^*mc_3}(Q_1+Q_2)$	$Q_2+Q_4) = \frac{-\Delta Hrx}{Cpc^*mc_3}C_2$	$\frac{-\Delta Hrx}{Cpc^*mc_3}C_{21}$	C	$\frac{-\Delta Hrx}{Cpc^*mc_3}C_{21}$
0	0	0	0	$\frac{-\Delta Hrx}{Cpc^*mc_3}(C_{22}-C_{21})$	$\frac{-\Delta Hrx}{Cpc^*mc_3}(C_{22}-C_{21})$	0	$\frac{-\Delta Hrx}{Cpc^{*}mc_{3}}(C_{22}-C_{21})$
0	0	0	0	$\frac{-\Delta Hrx}{Cpc^*mc_3}(C_{23}{-}C_{22})$	$\frac{-\Delta Hrx}{Cpc^*mc_3}(C_{23}-C_{22})$	0	$\frac{-\Delta Hrx}{Cpc^*mc_3}(C_{23}-C_{22}) \begin{bmatrix} T_F \\ T_F \end{bmatrix}$
0	0	0	0	$\frac{-\Delta Hrx}{Cpc^*mc_3}(C_{24}{-}C_{23})$	$\frac{-\Delta Hrx}{Cpc^*mc_3}(C_{24}-C_{23})$	0	$\frac{-\Delta Hrx}{Cpc^*mc_3}(C_{24}-C_{23}) \begin{vmatrix} T_{20} \\ C_F \end{vmatrix}$
0	0	0	0	$\frac{-\Delta Hrx}{Cpc^*mc_3}(C_{25}-C_{24})$	$\frac{-\Delta Hrx}{Cpc^*mc_3}(C_{25}-C_{24})$	0	$\frac{-\Delta Hrx}{Cpc^*mc_3}(C_{25}-C_{24})$
0	0	0	0	$\frac{-\Delta Hrx}{Cpc^*mc_3}(C_{26}-C_{25})$	$\frac{-\Delta Hrx}{Cpc^*mc_3}(C_{26}-C_{25})$	0	$\frac{-\Delta Hrx}{Cpc^*mc_3}(C_{26}-C_{25}) \begin{vmatrix} Q_1 \\ Q_2 \end{vmatrix}$
0	0	0	0	$\frac{-\Delta Hrx}{Cpc^{*}mc_{3}}(C_{27}-C_{26})$	$\frac{-\Delta Hrx}{Cpc^{*}mc_{3}}(C_{27}-C_{26})$	0	$\frac{-\Delta Hrx}{Cpc^*mc_3}(C_{27}-C_{26}) \begin{vmatrix} z_2 \\ Q_3 \end{vmatrix}$
0	0	0	0	$\frac{-\Delta Hrx}{Cpc^{*}mc_{3}}(C_{28}-C_{27})$	$\frac{-\Delta Hrx}{Cpc^{*}mc_{3}}(C_{28}-C_{27})$	0	$\frac{-\Delta Hrx}{Cpc^*mc_3}(C_{28}-C_{27}) Q_4$
0	0	0	0	$\frac{-\Delta Hrx}{Cpc^{*}mc_{3}}(C_{29}-C_{28})$	$\frac{-\Delta Hrx}{Cpc^*mc_3}(C_{29}-C_{28})$	0	$\frac{-\Delta Hrx}{Cpc^{*}mc_{3}}(C_{29}-C_{28})$
0	0	0	0	$\frac{-\Delta Hrx}{Cpc^{*}mc_{3}}(C_{30}-C_{29})$	$\frac{-\Delta Hrx}{Cpc^*mc_3}(C_{30}-C_{29})$	0	$\frac{-\Delta Hrx}{Cpc^*mc_3}(C_{23}-C_{29})$

$\alpha_1 = \frac{(Q_{B1})Cpg}{mc_1Cpc}$	
$\alpha_2 = \frac{(Q_{B2})Cpg}{mc_2Cpc}$	
$\alpha_3 = \frac{(Q_{B3})Cpg}{mc_3Cpc}$,
$\beta_1 = \frac{(Q_1)Cpg}{mc_1Cpc}$	
$\beta_2 = \frac{(Q_2)Cpg}{mc_2Cpc}$	
$\beta_3 = \frac{(Q_3)Cpg}{mc_3Cpc}$,
$\delta_1 = \frac{(Q_4)Cpg}{mc_1Cpc}$	
$\delta_2 = \frac{(Q_{B1})Cpg}{mc_2Cpc}$	
$\delta_3 = \frac{(Q_{B2})Cpg}{mc_3Cpc}$,

 $\begin{array}{l} \mathcal{Q}_{B1} = \mathcal{Q}_1 + \mathcal{Q}_4 \\ \\ \mathcal{Q}_{B2} = \mathcal{Q}_1 + \mathcal{Q}_2 + \mathcal{Q}_4 \\ \\ \mathcal{Q}_{B3} = \mathcal{Q}_1 + \mathcal{Q}_2 + \mathcal{Q}_3 + \mathcal{Q}_4 \end{array}$